

12526

03 Hours / 80 Marks

20112

Seat No.

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- Instructions –
- (1) All Questions are Compulsory.
 - (2) Answer each next main Question on a new page.
 - (3) Illustrate your answers with neat sketches wherever necessary.
 - (4) Figures to the right indicate full marks.
 - (5) Mobile Phone, Pager and any other Electronic Communication devices are not permissible in Examination Hall.

Marks

1. **Attempt any SIX of the following :** **30**
- a) Classify antibiotics based on their chemical structure. Draw the chemical structure of amoxicillin and state its brand names. (Any 2 brand names)
 - b) What are antimalarial drugs? Classify them with an example. Provide the chemical structure and uses of chloroquine.
 - c) Define and give two examples and their use for each of the following:
 - i) Cathartics
 - ii) Antimicrobial agents
 - d) Draw the chemical structure of phenytoin. Write its IUPAC name, therapeutic uses, formulations, and brand names.
 - e) Define and classify sympathomimetic agents with examples. Draw the chemical structure of naphazoline and give its uses.
 - f) What is volumetric analysis? List the apparatus used in this technique. Describe any one type of volumetric analysis.
 - g) Explain the principle and reaction involved in the limit test for arsenic. Draw a diagram of the Gutzeit apparatus.
2. **Attempt any TEN of the following :** **30**
- a) State the uses of chloramphenicol, azithromycin, and doxycycline.
 - b) Enlist any four sources of impurities and describe one of them in detail.
 - c) List two drugs from each of the following classes –
 - i) Antipsychotics
 - ii) Anticonvulsants
 - iii) Anti-depressants
 - d) Explain the principle of gravimetric analysis and describe the general method used in this technique.
 - e) Draw the structure of sulfacetamide. Write its uses and popular brand names (any 2 brand names)
 - f) Classify antineoplastic agents. Draw the structure of fluorouracil.
 - g) Draw the chemical structure of aspirin, ibuprofen & paracetamol.
 - h) Classify hypoglycemic agents with examples.
 - i) Draw the structure of Furosemide and write its chemical name. Give its uses.
 - j) Define and classify antihypertensive agents.
 - k) What are adrenergic antagonists? Classify them.

P.T.O.

3. Attempt ALL of the following :

- a) Lasix is a popular brand name for _____.
 - b) The turbidity observed in the limit test for sulfate is due to the formation of:
 - i) Barium sulfate
 - ii) Silver chloride
 - iii) Lead sulfate
 - iv) Calcium carbonate
 - c) Name any one barbiturate used as anaesthetic drug.
 - d) Draw structure of cyclophosphamide.
 - e) An analgesic containing para-aminophenol is _____.
 - f) Furan is:
 - i) A sulfur-containing five-membered fully unsaturated heterocyclic ring.
 - ii) A sulfur-containing five-membered saturated heterocyclic ring.
 - iii) An oxygen-containing five-membered fully unsaturated heterocyclic ring.
 - iv) An oxygen-containing five-membered saturated heterocyclic ring.
 - g) Write the uses of Atropine sulfate.
 - h) What is the prefix commonly used for oxygen in the nomenclature of heterocyclic compounds?
 - i) True or False: Sodium bicarbonate is an inorganic topical agent is primarily used for its antiseptic properties?
 - j) The ring system present in Imipramine is:
 - i) Dibenzazepine
 - ii) Benzodiazepine
 - iii) Phenothiazine
 - iv) Piperidine
 - k) Name any one drug used in the management of angina.
 - l) Define Haematinics.
 - m) The ring system present in the chemical structure of captopril is
 - i) Pyrrolidine
 - ii) Pyrrole
 - iii) Pyrazole
 - iv) Imidazole
 - n) Name any two tetracyclines.
 - o) Write any two uses of Pilocarpine.
 - p) Write any one use of Morphine and Diclofenac.
 - q) True or False: Imidazolidine is an unsaturated heterocyclic compound.
 - r) Sulfamethoxazole is often combined with another antibiotic to enhance its antibacterial effects. What is the common partner drug in this combination?
 - s) Draw structure of Quinoline heterocycle.
 - t) Define Limit Tests.
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