

MAHARASHTRA STATE BOARD OF TECHNICAL EDUCATION (Autonomous)

(ISO/IEC - 27001 - 2005 Certified)

#### SUMMER-18 EXAMINATION Model Answer

Subject: Heat Transfer Operation

Subject code: 17560

Page **1** of **23** 

#### Important Instructions to examiners:

- 1) The answers should be examined by key words and not as word-to-word as given in the model answer scheme.
- 2) The model answer and the answer written by candidate may vary but the examiner may try to assess the understanding level of the candidate.
- 3) The language errors such as grammatical, spelling errors should not be given more Importance (Not applicable for subject English and Communication Skills.
- 4) While assessing figures, examiner may give credit for principal components indicated in the figure. The figures drawn by candidate and model answer may vary. The examiner may give credit for any equivalent figure drawn.
- 5) Credits may be given step wise for numerical problems. In some cases, the assumed constant values may vary and there may be some difference in the candidate's answers and model answer.
- 6) In case of some questions credit may be given by judgement on part of examiner of relevant answer based on candidate's understanding.
- 7) For programming language papers, credit may be given to any other program based on equivalent concept.



| Q No.  | Answer   | marks  |
|--------|--|--------|
| 1 A    | Any three  | 12     |
| 1A-(i) | <b>Thermal conductivity</b> : It is the ability of measure of the substance to conduct   |        |
|        | heat. It is the quantity of heat passing through a material of a unit thickness          | 2      |
|        | with a unit heat flow area in unit time when a unit temperature difference is            |        |
|        | maintained across the opposite faces of the material.                                    |        |
|        | From Fourier's law   |        |
|        | Q = -kA(dT/dx)   | 1      |
|        | Or $k = Q.dx/(A.dT)$   |        |
|        | Substituting the units   |        |
|        | $k = W.m/(m^2.K)$  | 1      |
|        | = W/mK or J/(s.m.K)  |        |
| 1A-    | Film heat transfer coefficient: Film heat transfer coefficient h is defined as           | 2      |
| (ii)   | the quantity of heat transferred in unit time through unit area at a temperature         |        |
|        | difference of $1^0$ between the surface and surrounding.                                 |        |
|        | $1/U_o = 1/h_o + 1/h_i(D_o/D_i) + x_w/k(D_o/D_w) + R_d$                                  | 2<br>2 |
|        | $1/U_i = 1/h_i + 1/h_o(D_i/D_o) + x_w/k(D_i/D_w) + R_d$                                  |        |
| 1A-    | Stefan- Boltzman law:  |        |
| (iii)  | It states that the total energy emitted (emissive power) per unit area per unit          | 2      |
|        | time by a black body is proportional to fourth power of its absolute                     |        |
|        | temperature.   |        |
|        | $W_b \alpha T^4$   |        |
|        | Or $W_b = \sigma T^4$  | -      |
|        | Where $W_b = total$ energy emitted (emissive power) by a black body                      |        |
|        | $\sigma = \text{Stefan Boltzman constant} = 5.67 \times 10^{-8} \text{ W/m}^2 \text{ K}$ | 1      |



|        | T = absolute temperature                                      |        |
|--------|---|--------|
| 1A-    | Classification of shell and tube heat exchanger:              | 1 mark |
| (iv)   | 1. Fixed tube heat exchanger                                  | each   |
|        | 2. Floating head heat exchanger                               |        |
|        | 3. U- tube type heat exchanger                                |        |
|        | 4. Kettle/ Reboiler type heat exchanger                       |        |
| 1.B    | Any one   |        |
| 1B-(i) | Basis: 1 m length   |        |
|        | $r_1 = 0.0525m \ r_2 = 0.0575m$                               |        |
|        | $r_L = (r_2 - r_1) / \ln(r_2/r_1) = 0.055m$                   |        |
|        | $A_{L1} = 2\pi r_L L = 0.3452 m^2$                            |        |
|        | $K_1 = 43.03 \text{ W/mK}$                                    |        |
|        | $\mathbf{R}_{1}=\mathbf{B}_{1}/\mathbf{K}_{1}\mathbf{A}_{L1}$ |        |
|        | = 0.005/43.03 * 0.3452  |        |
|        | $= 3.37*10^{-4} \text{ K/W}$                                  |        |
|        | $r_2 = 0.0575m \ r_3 = 0.1075m$                               |        |
|        | $r_L = (r_3 - r_2) / \ln(r_3/r_2) = 0.08m$                    |        |
|        | $A_{L2} = 2\pi r_L L = 0.5018 m^2$                            |        |
|        | $K_2 = 0.07 \text{ W/mK}$                                     |        |
|        | $R_2 = B_2 / K_2 A_{L2}$                                      |        |
|        | $= 0.05/0.07* \ 0.5018$                                       |        |
|        | = 1.423 K/W   |        |
|        | $R= R_1 + R_2$  |        |
|        | = 1.4237 K/W  |        |
|        | Temp.drop $\Delta T$ = 120 K                                  |        |
|        | Heat loss $O = \Delta T / R$                                  |        |



|      | = <b>84.29</b> W   |   |
|------|--|---|
|      | $Q = (T_1 - T_2) / R_1$ where $T_2$ is the temperature at interface                | ] |
|      | $84.29 = (423 - T_2) / 3.37 * 10^{-4}$   |   |
|      | $T_2 = 422.97 K$   | ] |
| 1B-  | Methods of increasing the economy of an evaporator:                                |   |
| (ii) | 1. Using multiple effect evaporator  | 2 |
|      | 2. Vapour recompression  |   |
|      | A. Multiple effect evaporation: In this system, evaporators are arranged           |   |
|      | in series so that the vapour produced in first effect is fed to the steam          |   |
|      | chest of second effect as heating medium in which boiling takes place at           |   |
|      | low pressure and temperature and so on.  |   |
|      | B. Methods of increasing economy by vapour recompression methods are:              |   |
|      | 1. Mechanical recompression  |   |
|      | 2. Thermal recompression   |   |
|      | <b>Thermal recompression:</b> To increase the economy of single effect evaporator, |   |
|      | the principle of thermal recompression is used. Here the vapour from the           |   |
|      | evaporator is compressed to increase its temperature so that it will condense at a | 2 |
|      | temperature higher enough to permit its use as heating media in the same           |   |
|      | evaporator. In this method, vapour is compressed by means of jet ejectors. Here    |   |
|      | the high pressure steam is used to draw and compress the major part of vapours     |   |
|      | from the evaporator, while the remaining part of vapours is separately             |   |
|      | condensed for compensating motive steam added.                                     |   |
|      |  |   |



| : Heat ' | Transfer Operation Subject code: 17560   | Page <b>6</b> of <b>23</b> |
|----------|--|----------------------------|
|          | Total cost   | 1                          |
|          | Optimum Thickness Of Insulation  |                            |
| 2-b      | Fourier's law of conduction:   |                            |
|          | It states that the rate of heat flow across an isothermal surface is proportional to   | 2                          |
|          | the temperature gradient at the surface.   |                            |
|          | $\frac{dQ}{dA} = -k\frac{\delta T}{\delta n}$  | 1                          |
|          | Q- rate of heat transfer   |                            |
|          | A- Area perpendicular to heat flow   | 1                          |
|          | k- Thermal conductivity  |                            |
|          | T- Temperature   |                            |
| 2-c      | Kirchhoff's Law :  |                            |
|          | Consider that the two bodies are kept into a furnace held at constant                  | 2                          |
|          | temperature of T K. Assume that, of the two bodies one is a black body& the            |                            |
|          | other is a non-black body i.e. the body having 'a' value less than one. Both the       |                            |
|          | bodies will eventually attain the temperature of T K & the bodies neither              |                            |
|          | become hotter nor cooler than the furnace. At this condition of thermal                |                            |
|          | equilibrium, each body absorbs and emits thermal radiation at the same rate.           |                            |
|          | The rate of absorption & emission for the black body will be different from that       |                            |
|          | of he non-black body.  |                            |
|          | Let the area of non-black body be $A_1$ and $A_2$ respectively. Let 'I' be the rate at |                            |



| ct: Heat Transfer Operation                       | Subject code: 17560                              | Page <b>7</b> of <b>23</b> |
|---|--|----------------------------|
| which radiation falling on bodies pe              | er unit area and $E_1$ and $E_2$ be the emissive |                            |
| powers ( emissive power is the tota               | al quantity of radiant energy emitted by a       |                            |
| body per unit area per unit time)of no            | on-black & black body respectively.              |                            |
| At thermal equilibrium, absorption                | and emission rates are equal, thus,              | 2                          |
| $Ia_1 A_1 = A_1 E_1$                              | (1.1)  |                            |
| $\therefore$ Ia <sub>1</sub> = E <sub>1</sub>     | (1.2)  |                            |
| And $Ia_b A_2 = A_2 E_b$                          | (1.3)  |                            |
| $Ia_b = E_b$                                      | (1.4)  |                            |
| From equation (1.1) and (1.4).we get              |  |                            |
| $\frac{E1}{a1} = \frac{Eb}{ab}$                   | (1.5)  |                            |
| Where $a_{1,a_b}$ = absorptivity of non-bla       | ck & black bodies respectively.                  |                            |
| If we introduce a second body (ne                 | on-black) then for the second non-black          |                            |
| body,we have :                                    |  |                            |
| $I A_3 a_2 = E_2 A_3$                             | 3(1.6)   |                            |
| $\therefore Ia_2 = E_2$                           | (1.7)  |                            |
| Where $a_1 = E_2$ are the absorptivity and        | d emissive power of the second non-black         |                            |
| body.   |  |                            |
| Combining equations (1.2),(1.4) an                | .d(1.7) we get,                                  |                            |
| $\frac{E1}{a1} = \frac{E2}{a2} = \frac{E3}{a3} =$ | E <sub>b</sub> (1.8)                             |                            |
| 2-d Application of finned tube hea                | at exchanger: When the heat transfer             | 2                          |
| coefficient of one of the process fluid           | ds is very low as compared to the other, the     |                            |
| overall heat transfer coefficient be              | comes approximately equal to the lower           |                            |
| coefficient. This reduces the capacity            | y per unit area of the heat transfer surface     |                            |







### SUMMER-18 EXAMINATION Model Answer

Subject: Heat Transfer Operation

Subject code: 17560

Page **9** of **23** 

|   | $A = 19.95 m^2$ |                                     |                          | 1        |
|---|-----------------|-------------------------------------|--------------------------|----------|
| ) | Dropwise and f  | ilmwise condensation:               |                          | 2 marks  |
|   | Points          | Dropwise condensation               | Filmwise                 | each for |
|   |                 |                                     | condensation             | any 4    |
|   | mechanism       | In case of drop-wise condensation   | In case of film-wise     |          |
|   |                 | the condensate (condensed liquid)   | condensation the         |          |
|   |                 | does not wet the surface and        | condensed liquid wets    |          |
|   |                 | collects to grow for a while and    | the surface and forms a  |          |
|   |                 | then fall from the surface, leaving | continuous film of       |          |
|   |                 | bare metal surface for further      | condensate through       |          |
|   |                 | condensation.                       | which heat transfer      |          |
|   |                 |                                     | takes place. This        |          |
|   |                 |                                     | condensate flows down    |          |
|   |                 |                                     | due to action of gravity |          |
|   | Heat transfer   | Heat transfer coefficient are very  | Heat transfer            |          |
|   | coefficient     | high in case of drop-wise           | coefficients are         |          |
|   |                 | condensation since the heat does    | relatively very low in   |          |
|   |                 | not have to flow through film by    | case of film-wise        |          |
|   |                 | conduction                          | condensation since the   |          |
|   |                 |                                     | heat does have to flow   |          |
|   |                 |                                     | through film by          |          |
|   |                 |                                     | conduction               |          |
|   | Surface type    | Oily or greasy surfaces seem to     | Smooth, clean surfaces   |          |
|   |                 | tend towards drop-wise              | seem to tend towards     |          |
|   |                 | condensation                        | film-wise condensation   |          |
|   | Stability       | Drop-wise condensation is very      | Film-wise                |          |
|   |                 |                                     |                          |          |



|        |  | difficult to achieve a  | nd unstable    | condensation is easily                |          |
|--------|--|---|----------------|---------------------------------------|----------|
|        |  |   |                | obtainable and stable                 |          |
|        | equations  | If the students write e   | equations for  | If the students write                 |          |
|        |  | film coefficients on v  | vertical and   | equations for film                    |          |
|        |  | horizontal surfaces m   | arks should    | coefficients on vertical              |          |
|        |  | be given  |                | and horizontal surfaces               |          |
|        |  |   |                | marks should be given                 |          |
| 3-c    | Comparison of  | square pitch and trian  | gular pitch(a  | ny 4)                                 | 1.5 mark |
|        | Sq   | uare pitch  | Tr             | iangular pitch                        | each     |
|        | Permits externa  | al cleaning of the tubes  | Difficult to c | lean                                  |          |
|        | Causes low pre   | essure drop on the shell  | Causes more    | pressure drop                         |          |
|        | side fluid   |   |                |                                       |          |
|        | Less no.   | of tubes can be   | Larger no.     | of tubes can be                       |          |
|        | accommodated   | than with triangular  | accommodat     | ed in a given shell                   |          |
|        | pitch  | d   |                |                                       |          |
|        | Creates c  | Creates comparatively less  |                | Creates large turbulence in the shell |          |
|        | turbulence   | turbulence  |                | side fluid                            |          |
|        | Can be used for dirty fluids also Used for clean fluid |   |                |                                       |          |
|        | Use of baffle:   |   |                |                                       |          |
|        | 1. To incre  | 1. To increase the rate of heat transfer by increasing the velocity and |                |                                       |          |
|        | turbulen   | turbulence of the shell side fluid.                                     |                |                                       | each     |
|        | 2. Structura   | al support for the tubes a  | and dampers ag | gainst vibration.                     |          |
| 4 A    | Any three  |   |                |                                       | 12       |
| 4A-(i) | Heat transfer t  | hrough single flat furn   | ace wall :     |                                       | 2        |
| 1      |  |   |                | 1 1 1 77 0 1 0                        |          |



| t: Heat     | Transfer Operation  | Subject code:   | 17560  | Page <b>11</b> of <b>23</b>       |
|-------------|---|---|--|-----------------------------------|
|             | independent of temperature & heat losses t  | to atmosphere is negli  | gible. Hot face  | e                                 |
|             | is at a temperature $T_1$ & cold face is at a te  | emperature T <sub>2</sub> . The dir   | ection of heat   |                                   |
|             | flow is perpendicular to the wall & T varie   | es in direction of X-ax   | is.  |                                   |
|             | T face dt cold face<br>x=0 KH dx  | e   |  |                                   |
|             | At Steady State, there can be neither accum<br>a plane wall &Q is constant along heat flow<br>requires that the differential eqn is integrat  | mulation nor depletion<br>w. The ordinary use of<br>ted over entire path fro  | of heat within<br>f Fourier's Latom $x = 0, x = x$                                   | n<br>w                            |
|             | At Steady State, there can be neither accum<br>a plane wall &Q is constant along heat flow<br>requires that the differential eqn is integrat<br>$\therefore Q = -K A$   | mulation nor depletion<br>w. The ordinary use of<br>ted over entire path fro<br>AdT/dx  | of heat within<br>f Fourier's Latom $x = 0, x = x$                                   | n<br>w                            |
|             | At Steady State, there can be neither accum<br>a plane wall &Q is constant along heat flow<br>requires that the differential eqn is integrat<br>$\therefore Q = -K A$<br>Q dx = -K A  | mulation nor depletion<br>w. The ordinary use of<br>ted over entire path fro<br>AdT/dx<br>A dT  | of heat within<br>f Fourier's Lat<br>om x = 0,x = x                                  | n<br>w<br><br>2                   |
|             | At Steady State, there can be neither accum<br>a plane wall &Q is constant along heat flow<br>requires that the differential eqn is integrat<br>$\therefore Q = -K A$<br>Q dx = -K A<br>OR  | mulation nor depletion<br>w. The ordinary use of<br>ted over entire path fro<br>AdT/dx<br>A dT  | of heat within<br>f Fourier's Lat<br>om x = 0,x = x                                  | n<br>w<br><br>2                   |
|             | At Steady State, there can be neither accur<br>a plane wall &Q is constant along heat flow<br>requires that the differential eqn is integrat<br>$\therefore Q = -K A$<br>Q dx = -K A<br>OR<br>$Q_0 \int^x dx = -K A$  | mulation nor depletion<br>w. The ordinary use of<br>ted over entire path fro<br>AdT/dx<br>A dT<br>A T1 <sup>JT2</sup> .dt   | of heat within<br>f Fourier's Latom x = 0,x = x                                      | n<br>w<br><br>2                   |
|             | At Steady State, there can be neither accur<br>a plane wall &Q is constant along heat flow<br>requires that the differential eqn is integrat<br>$\therefore Q = -K A$<br>Q dx = -K A<br>OR<br>$Q_0 \int^x dx = -K A$<br>Q.x = -K A  | mulation nor depletion<br>w. The ordinary use of<br>ted over entire path fro<br>AdT/dx<br>A dT<br>A $_{T1}\int^{T2}$ .dt<br>A $(T_2 - T_1)$   | of heat within<br>f Fourier's Lay<br>om x = 0,x = x                                  | n<br>w<br><br>2                   |
|             | At Steady State, there can be neither accur<br>a plane wall &Q is constant along heat flow<br>requires that the differential eqn is integrat<br>$\therefore Q = -K A$<br>Q dx = -K A<br>OR<br>$Q_0 \int^x dx = -K A$<br>Q.x = -K A<br>OR  | mulation nor depletion<br>w. The ordinary use of<br>ted over entire path fro<br>AdT/dx<br>A dT<br>A $_{T1}\int^{T2}$ .dt<br>A $(T_2 - T_1)$   | of heat within<br>f Fourier's Lav<br>om x = 0,x = x                                  | n<br>w<br><br>2                   |
|             | At Steady State, there can be neither accur<br>a plane wall &Q is constant along heat flow<br>requires that the differential eqn is integrat<br>$\therefore Q = -K A$<br>Q dx = -K A<br>Q dx = -K A<br>Q.x = -K A<br>Q.x = -K A<br>Q.x = -K A   | mulation nor depletion<br>w. The ordinary use of<br>ted over entire path fro<br>AdT/dx<br>A dT<br>A $_{T1}\int^{T2}$ .dt<br>A $(T_2 - T_1)$   | of heat within<br>f Fourier's Lav<br>om x = 0,x = x                                  | n<br>w<br><br>2                   |
| 4A-         | At Steady State, there can be neither accur<br>a plane wall &Q is constant along heat flow<br>requires that the differential eqn is integrat<br>$\therefore Q = -K A$<br>Q dx = -K A<br>Q dx = -K A<br>Q x = -K A   | mulation nor depletion<br>w. The ordinary use of<br>ted over entire path fro<br>AdT/dx<br>A dT<br>A $_{T1}\int^{T2}$ .dt<br>A $(T_2 - T_1)$   | of heat within<br>f Fourier's Lay<br>om x = 0,x = x                                  | n<br>w<br><br>2<br>1 mark         |
| 4A-<br>(ii) | At Steady State, there can be neither accur<br>a plane wall &Q is constant along heat flow<br>requires that the differential eqn is integrat<br>$\therefore Q = -K A$<br>Q dx = -K A<br>Q dx = -K A<br>Q x = -K A   | mulation nor depletion<br>w. The ordinary use of<br>ted over entire path fro<br>AdT/dx<br>A dT<br>A $_{T1}\int^{T2}$ .dt<br>A $(T_2 - T_1)$<br>gements: (any 4)<br>Backward fee                                 | of heat within<br>f Fourier's Lay<br>om x = 0,x = x                                  | n<br>w<br><br>2<br>1 mark<br>each |
| 4A-<br>(ii) | At Steady State, there can be neither accur<br>a plane wall &Q is constant along heat flow<br>requires that the differential eqn is integrat<br>$\therefore Q = -K A$<br>Q dx = -K A<br>Q dx = -K A<br>Q x = -K A | mulation nor depletion<br>w. The ordinary use of<br>ted over entire path fro<br>AdT/dx<br>A dT<br>A $_{T1}$ $\int^{T2}$ .dt<br>A $(T_2 - T_1)$<br>gements: (any 4)<br>Backward fee<br>Flow of solution to be of | of heat within<br>f Fourier's Lay<br>om $x = 0, x = x$<br><b>d</b><br>concentrated i | n<br>w<br><br>2<br>1 mark<br>each |







MAHARASHTRA STATE BOARD OF TECHNICAL EDUCATION (Autonomous)

(ISO/IEC - 27001 - 2005 Certified)





| bject: Heat T | ransfer Operation Subject code: 17560  | Page <b>14</b> of <b>23</b> |
|---------------|--|-----------------------------|
|               | conductivity. Graphite being soft, these exchangers are made in cubic or             |                             |
|               | cylindrical blocks. In cubic exchangers, parallel holes are drilled in a solid cube  |                             |
|               | such that parallel holes of a particular row are at right angles to the holes of the |                             |
|               | row above & below. Headers bolted to the opposite sides of the vertical faces of     |                             |
|               | the cube provide the flow of process fluid through the block. The headers            |                             |
|               | located on the remaining vertical faces direct the service fluid through the         |                             |
|               | exchanger in a cross flow.   |                             |
|               |  | 2                           |
| 4 B           | Any one  | 6                           |
| 4B-(i)        | let area = $1 \text{ m}^2$   |                             |
|               | Thermal resistance o fire brick = $x_1/k_1 A$  | 1                           |
|               | $R_1 = 0.23/1.21 x 1 = 0.190 k/w$  |                             |
|               | Similarly $R_2 = x_2/k_2 A = 0.075/0.121x1 = 0.62 k/w$                               |                             |
|               | $R_3 = x_3/k_3 A = 0.089/0.865x1 = 0.103 k/w$  |                             |
|               | $\mathbf{R} = \mathbf{R}_1 + \mathbf{R}_2 + \mathbf{R}_3$                            | 2                           |
|               | R = 0.913  k/w   |                             |
|               | The heat loss per unit area is $Q = \Delta T/R$                                      |                             |
|               | $\Delta T = (1073 - 333) = 740 \text{ k}$  |                             |
|               | Q=740/0.913 = 810.51W  | 1                           |
|               | Q= $(T_1-T_2)$ / $R_1$ where $T_2$ is the temp at interface between fire brick and   |                             |
|               | insulating brick   |                             |
| 1             |  | 1                           |







| et: Heat ' | Transfer Operation Subject code: 17560   | Page <b>16</b> of <b>23</b> |  |
|------------|--|-----------------------------|--|
|            | pipe connecting the vapour space to the bottom of the exchanger is             | 2                           |  |
|            | provided for natural circulation of a unvapourised liquid. It is provided with |                             |  |
|            | inlet connection for feed, steam and outlet connections for vapour, thick      |                             |  |
|            | liquor, condensate etc.  |                             |  |
|            | Working:   |                             |  |
|            | In this evaporator feed enters the bottom of the tubes, gates heated by the    |                             |  |
|            | condensing steam, starts to boil part way up the tubes and the mixture of      |                             |  |
|            | vap. and liquid comes out from the top of the tubes and finally impinges at    |                             |  |
|            | high velocity on a deflector. The deflector acts both as a primary separator   |                             |  |
|            | and foam breaker. The separated liquid enters the bottom of the exchanger      |                             |  |
|            | and parts of this liquid is taken out as a product.                            | 2                           |  |
|            | This type of evaporator is widely used for handling of foamy, frothy           |                             |  |
|            | liquids.   |                             |  |
|            | It is typically used for the production of condensed milk and concentrating    |                             |  |
|            | black liquor in the pulp and paper industry.                                   |                             |  |
| 5          | Any two  | 16                          |  |
| 5-a        | To derive Q=UA ΔT <sub>lm</sub>  |                             |  |
|            | Assumptions:   |                             |  |
|            | 1. Overall coefficient U is constant throughout the exchanger                  | 2                           |  |
|            | 2. Specific heats of hot and cold fluids are constant                          |                             |  |
|            | 3. Heat flow to and from the ambient is negligible                             |                             |  |
|            | 4. Flow is steady and may be parallel or counter current type                  |                             |  |
|            | 5. Temperatures of both the fluids are uniform over a given cross section      |                             |  |
|            | and may be represented by their bulk temperature.                              |                             |  |
|            |  |                             |  |
|            |  |                             |  |
|            |  |                             |  |





| oject: Heat | Transfer Operation S  | ubject code:              | 17560        | Page <b>18</b> of <b>23</b> |
|-------------|---|---------------------------|--------------|-----------------------------|
|             | $\int_{\Delta Ti} d(\Delta T) / \Delta T = - (1/(mh Cph) + 1/(mc Cpc)) U$ | $B \int_0^L dx$           |              |                             |
|             | $\ln (\Delta Te/\Delta Ti) = - (1/(mh Cph) + 1/(mc Cpc)) UA$              |                           |              | -(6)                        |
|             | where $\Delta Te = T_{he} - T_{ce}$                                       |                           |              | 1                           |
|             | $\Delta Ti = T_{hi} - T_{ci}$   |                           |              |                             |
|             | Now if q is the total rate of heat transfer in the heat                   | exchanger, t              | then         |                             |
|             | $q = m_h C p_h (T_{hi} - T_{he})$ (7)                                     |                           |              |                             |
|             | = mc Cpc (T <sub>ce</sub> - T <sub>ci</sub> )(8)                          |                           |              |                             |
|             | Substituting equations (7) and (8) into equation (6),                     | ,                         |              | 1                           |
|             | $\ln (\Delta Te/\Delta Ti) = -1/q[ (T_{hi}-T_{he}) + (T_{ce}-T_{ci})]U A$ |                           |              |                             |
|             | q= U A ( $\Delta$ Ti- $\Delta$ Te)/ ln ( $\Delta$ Ti/ $\Delta$ Te)        | (9)                       |              |                             |
|             | Equation (9) is the performance equation for a para                       | llel-flow hea             | ıt exchanger |                             |
|             | $Q = U A \Delta T lm$   |                           |              |                             |
|             | Where $\Delta Tlm = (\Delta Ti - \Delta Te) / ln (\Delta Ti / \Delta Te)$ |                           |              | 1                           |
| 5-b         | Material balance equation for single effect evapo                         | orator:                   |              |                             |
|             | Consider that the evaporator is fed with $m_f kg/h$ of                    | weak solutio              | n containing | $g w_1$                     |
|             | % solute & thick liquor is withdrawn at m' kg/h con                       | ntaining w <sub>2</sub> % | % solids by  | 1                           |
|             | weight. Let $m_v$ be the kg/h of water evaporated. The                    | en :                      |              |                             |
|             |   |                           |              |                             |
|             |   |                           |              |                             |



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| : Heat | Transfer Operation Subject code: 17560   | Page <b>20</b> of <b>2</b> |
|--------|--|----------------------------|
|        | Heat transfer to solution in evaporator by condensing steam (in absece of heat   |                            |
|        | losses) is utilised to heat the feed solution from Tf to T and for vaporisation of   | 1                          |
|        | water from solution.   |                            |
|        | Qs = Q   |                            |
|        | $= m_f \operatorname{Cpf} (T - T_f) + (m_f - m') \lambda_v \dots \dots (vii)$  |                            |
|        | $m_s$ . $\lambda_s = mf \ Cpf \ (T - T_f) + (m_f - m') \lambda_v \dots (viii)$   |                            |
|        | where $Cp_f =$ specific heat of feed solution  |                            |
|        | $\lambda_v$ = latent heat of evaporation from thick liquor   |                            |
|        | For negligible boiling point rise $\lambda v = \lambda$  |                            |
|        | Where $\lambda$ =latent heat of vaporisation of water at pressure in the   |                            |
|        | Vapour space & can be read from steam tables.  | 1                          |
|        | Above equation (viii) becomes :  |                            |
|        | $m_s \lambda_s = m_f C p_f (T - T_f) + (m_f - m^2) \lambda(ix)$  |                            |
|        | $m_s \lambda_s = m_f  C p_f  (T - T_f) + m_v  \lambda(x)$  |                            |
| 5-c    | The Sider – Tate equation is   | 2                          |
|        | hi Di/k = 0.023 (NRe <sup>) 0.8</sup> (Npr) <sup>1/3</sup> ( $\mu/\mu w$ ) <sup>0.14</sup>                                     |                            |
|        | Substituting all the values in the equation we get   |                            |
|        | hi $(0.02)/0.25 = 0.023 \text{ x} (15745)^{0.8} (36)^{1/3} \text{ x} ((550 \text{ x} 10^{-6})/(900 \text{ x} 10^{-6}))^{0.14}$ | 2                          |
|        | hi (0.02)/0.25 = 0.023 x 2278.84 x 3.3 x 0.933   |                            |
|        | hi (0.02)/0.25 = 161.37  | 2                          |
|        | hi= 2017   |                            |
|        | Inside heat transfer coefficient = 2017 W/m <sup>2</sup> .K  | 2                          |
| 6      | Any two  | 16                         |
| 6-a    | Dimensional Analysis :   | 2                          |
|        | It is a method of correlating a number of variables into a single equation   |                            |
|        | expressing an effect.  |                            |



| ubject: Heat Transfer Operation                                   | Subject code: 17560                          | Page <b>21</b> of <b>23</b> |
|---|--|-----------------------------|
| Dimensional analysis is a method o                                | f reducing the number of variables required  | L L                         |
| to describe a given physical situatio                             | n by making use of the information implied   | b                           |
| by the units of the physical quantiti                             | es involved. It is also known as the "theory | y                           |
| of similarity".   |  |                             |
| Dittus – Bolter equation:   |  |                             |
| $hD/k = 0.023[(Du\rho/\mu)^{0.8}(Cp \mu/k)^{a}$                   |  | 1                           |
| where $a = 0.4$ for heating                                       |  |                             |
| a=0.3 for cooling.  |  |                             |
| where h= film heat transfer coeffici                              | ent  |                             |
| D= diameter of pipe line  |  |                             |
| $\mu$ = viscosity of the liquid                                   |  | 2                           |
| $\mu$ w= viscosity of the liquid at the wa                        | all surface temp                             |                             |
| Cp= specific heat of the liquid                                   |  |                             |
| L= length of pipe.  |  |                             |
| k= thermal conductivity   |  |                             |
| u= velocity of flow   |  |                             |
| The Sider – Tate equation is                                      |  |                             |
| hi Di/k = 0.023 (NRe) <sup>0.8</sup> (Npr) <sup>1/3</sup> ( $\mu$ | $(\mu w)^{0.14}$                             | 1                           |
| where $h = film$ heat transfer coeffici                           | ent  |                             |
| D= diameter of pipe line  |  |                             |
| $\mu$ = viscosity of the liquid                                   |  |                             |
| $\mu$ w= viscosity of the liquid at the wa                        | all surface temp                             |                             |
| Cp= specific heat of the liquid                                   |  | 2                           |
| L= length of pipe.  |  |                             |
| k= thermal conductivity   |  |                             |
| u= velocity of flow   |  |                             |
|   |  | 1                           |



| ct: Heat '   | Transfer Operation                  | Subject code: 17560  | Page <b>22</b> of <b>2</b> 3 |
|--------------|-------------------------------------|--|------------------------------|
| 6-b          | 328 K Cold fluid 358 K              | (t <sub>1</sub> ) 328 Cold fluid $\rightarrow$ 358 K (t <sub>2</sub> )   |                              |
|              | 578 K Thermic fluid 433 K $(T_1)$   | 433 K Thermic fluid 578  | К 2                          |
|              |                                     |  |                              |
|              | Co-current flow                     | Counter current flow   |                              |
|              | co current flow                     |  |                              |
|              | $\Delta T_1 =$                      | 578 - 328 = 250  K   |                              |
|              | $\Delta T_2 = -$                    | 433 – 358 = 75 К   | 1                            |
|              | $LMTD = \frac{\Delta T1}{\ln t}$    | $\frac{\Delta T2}{\left(\frac{\Delta T1}{\Delta T2}\right)} = \frac{250 - 75}{\ln\left(\frac{250}{75}\right)} = 145.35 \text{K}$   |                              |
|              | Total heat transferred $Q = U A L$  | MTD  | 1                            |
|              | = 700 * 5                           | 500 * 145.35   |                              |
|              | = 508732                            | 242.14 W or 50873.242 kW   |                              |
|              | counter current flow                |  | 1                            |
| $\Delta T_1$ |                                     | 433 – 328 = 105 K  | -                            |
|              | $\Delta T_2 =$                      | 578 – 358 = 220 K  | 1                            |
|              | $LMTD = \frac{\Delta T1}{\ln t}$    | $\frac{\Delta T2}{\left(\frac{\Delta T1}{\Delta T2}\right)} = \frac{105 - 220}{\ln\left(\frac{105}{220}\right)} = 155.48 \text{K}$ |                              |
|              | Total heat transferred $Q = U A L$  | MTD  | 1                            |
|              | = 700 * 5                           | 500 * 155.48   |                              |
|              | = 544163                            | 364.83 W or 54416.364 kW   |                              |
| 6-c          | Basis: 5000 kg/hr feed is fed to th | ne evaporator.   |                              |
|              | Material balance of solids:         |  |                              |



| Subject: Heat Transfer OperationSubject code: 17560                    | Page <b>23</b> of <b>23</b> |
|--|-----------------------------|
| Solids in feed= solids in the thick liquor                             | 1                           |
| 0.01x5000=0.4 x m'   |                             |
| m'=1250kg/h.   |                             |
| overall Material balance:  |                             |
| kg/h feed= $kg/h$ water evaporated + $kg/h$ thick liquor               | 1                           |
| water evaporated( $m_v$ )=5000-1750=3750kg/h                           |                             |
| Energy balance is  |                             |
| $m_s \ \lambda_s = m^* c_{pf} \ ^* (T\text{-}T_f) + m_v \ \ \lambda_v$ | 1                           |
| $m_s 2162 = 5000*4.187*(373-313) + 3750$ (2676-419)                    |                             |
| steam fed( $m_s$ )= 4495.77 kg/h                                       | 1                           |
| steam economy= kg/h water evaporated/kg/h steam consumed               |                             |
| = 3750/4495.77= <b>0.834</b>   | 1                           |
| $Q = U^*A^*\Delta T$   | 1                           |
| 4495.77*2162*1000/ 3600 = 1750 * A*(373-313)                           |                             |
| $A = 45.38 m^2$  | 2                           |